

## Four New Flavonol Glycosides from the Leaves of *Astragalus caprinus*

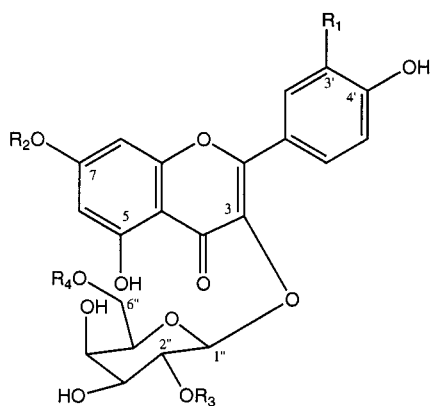
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Received June 29, 2001

Four new flavonol 3-*O*-glycosides were isolated from the leaves of *Astragalus caprinus*. Their structures were elucidated by spectroscopic methods as rhamnocitrin-3-*O*-{[3-hydroxy-3-methylglutaroyl(1→6)]-[β-D-apiofuranosyl(1→2)]}-β-D-galactopyranoside (**1**), rhamnetin-3-*O*-{[3-hydroxy-3-methylglutaroyl(1→6)]-[β-D-apiofuranosyl(1→2)]}-β-D-galactopyranoside (**2**), kaempferol-3-*O*-[β-D-xylopyranosyl(1→3)-α-L-rhamnopyranosyl(1→6)]-β-D-galactopyranoside (**3**), and quercetin-3-*O*-{[β-D-xylopyranosyl(1→3)-α-L-rhamnopyranosyl(1→6)]-[β-D-apiofuranosyl(1→2)]}-β-D-galactopyranoside (**4**).

*Astragalus caprinus* Maire (Fabaceae) is an endemic species of North Africa, the leaves of which are used as an antihemorrhoidal agent in Tunisian folk medicine. Our previous paper describing the structure of a new 3-*O*-tetraglycoside of kaempferol from this plant was the first phytochemical report on this species.<sup>1</sup> Herein we report the isolation and structure elucidation of four new flavonol glycosides. Their structures were elucidated by spectroscopic methods as rhamnocitrin-3-*O*-{[3-hydroxy-3-methylglutaroyl(1→6)]-[β-D-apiofuranosyl(1→2)]}-β-D-galactopyranoside (**1**), rhamnetin-3-*O*-{[3-hydroxy-3-methylglutaroyl(1→6)]-[β-D-apiofuranosyl(1→2)]}-β-D-galactopyranoside (**2**), kaempferol-3-*O*-[β-D-xylopyranosyl(1→3)-α-L-rhamnopyranosyl(1→6)]-β-D-galactopyranoside (**3**), and quercetin-3-*O*-{[β-D-xylopyranosyl(1→3)-α-L-rhamnopyranosyl(1→6)]-[β-D-apiofuranosyl(1→2)]}-β-D-galactopyranoside (**4**). Compounds **1–4** are new according to the literature.<sup>2–4</sup>



	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>
<b>1</b>	H	CH <sub>3</sub>	api	3-OH-3-CH <sub>3</sub> -glut
<b>2</b>	OH	CH <sub>3</sub>	api	3-OH-3-CH <sub>3</sub> -glut
<b>3</b>	H	H	H	-rha 3-xy
<b>4</b>	OH	H	api	-rha 3-xy

Compound **1** was isolated as an amorphous yellow powder. Its UV spectral properties indicated a 3,7-disubstituted kaempferol derivative.<sup>5,6</sup> Chromatography of the

acid hydrolysis products allowed the identification of 7-methylkaempferol (rhamnocitrin) and apiose and galactose. The HRFABMS (positive-ion mode) of **1** exhibited a pseudomolecular ion peak at  $m/z$  761.1908 [M + Na]<sup>+</sup> (calcd 761.1905), consistent with a molecular formula of C<sub>33</sub>H<sub>38</sub>O<sub>19</sub>. Its ESIMS (positive-ion mode) displayed a pseudomolecular ion peak [M + H]<sup>+</sup> at  $m/z$  739 and fragment ion peaks at  $m/z$  607 [M + H - 132]<sup>+</sup>, indicating the loss of apiose, and at  $m/z$  301 [M + H - 132 - 306]<sup>+</sup>, which was assigned to the additional loss of galactose (-162) linked to an unassigned moiety (-144). The peak at 1720 cm<sup>-1</sup> in the IR spectrum was suggestive of a saturated acyl moiety. The <sup>1</sup>H NMR spectrum of **1** exhibited a signal at δ 3.95 (3H, s), which correlated in the HSQC spectrum to a <sup>13</sup>C NMR signal at δ 56.5 and corresponded to the methoxy group at C-7 of the aglycon (Table 1).<sup>7</sup> Four aromatic doublets corresponded to H-6 [δ 6.30 (1H, *J* = 1.9 Hz)], *meta*-coupled with H-8 [δ 6.59 (1H, *J* = 1.9 Hz)] on the A ring, and to H-2', H-6' [δ 8.13 (2H, *J* = 8.8 Hz)], *ortho*-coupled with H-3', H-5' [δ 6.90 (2H, *J* = 8.8 Hz)]. Full identification of the aglycon was finally achieved by 2D NMR spectroscopy, which led to the rhamnocitrin structure. Two anomeric protons at δ 5.44 (1H, d, *J* = 7.5 Hz) and at δ 5.45 (1H, d, *J* = 1.6 Hz) were assigned to a β-galactopyranosyl (Gal) and a β-apiofuranosyl (Api) unit, respectively, by a HSQC-TOCSY experiment and coupling constant measurements. HMBC experiments showed correlations between Gal H-1 and rhamnocitrin C-3 (δ 135.0) and between Api H-1 and Gal C-2 (δ 76.8). The downfield chemical shifts of the Gal methylene protons and Gal C-6 (Table 1) suggested acylation of the galactose unit at C-6.<sup>8</sup> These protons were correlated in the HMBC spectrum with a carboxylic carbon at δ 172.2. The HSQC and HMBC spectra showed this carboxyl to belong to an aliphatic acyl moiety, where it was successively linked to a methylene group [δ<sub>C</sub> 46.0; δ<sub>H</sub> 2.35 (d, *J* = 15.1 Hz); δ<sub>H</sub> 2.42 (d, *J* = 15.1 Hz)], a central quaternary carbon (δ 70.6), another methylene group [δ<sub>C</sub> 46.3; δ<sub>H</sub> 2.36 (d, *J* = 14.9 Hz); δ<sub>H</sub> 2.46 (d, *J* = 14.9 Hz)], and a terminal carboxyl (δ 175.6). Additionally, the central quaternary carbon (C-3) showed a HMBC correlation with a methyl group at δ<sub>H</sub> 1.13 (3H, s) and δ<sub>C</sub> 27.9. Consequently, the aliphatic acyl moiety was identified as a 3-hydroxy-3-methylglutaroyl substituent. On the basis of all these data, **1** was identified as rhamnocitrin-3-*O*-{[3-hydroxy-3-methylglutaroyl(1→6)]-[β-D-apiofuranosyl(1→2)]}-β-D-galactopyranoside. Esterification with 3-hydroxy-3-

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**Table 1.**  $^{13}\text{C}$  and  $^1\text{H}$  NMR Spectral Data of Compounds **1** and **2** in  $\text{CD}_3\text{OD}$ 

position	<b>1</b>		<b>2</b>	
	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J/Hz)	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J/Hz)
aglycon				
2	158.8		158.8	
3	135.0		135.2	
4	179.6		179.5	
5	162.8		162.9	
6	99.0	6.30 d (1.9)	99.0	6.30 d (1.9)
7	167.2		167.2	
8	93.1	6.59 d (1.9)	93.0	6.58 d (1.9)
9	158.3		158.3	
10	106.7		106.7	
1'	122.8		123.1	
2'	132.4	8.13 d (8.8)	117.3	7.72 d (2.2)
3'	116.2	6.90 d (8.8)	149.9	
4'	161.6		146.0	
5'	116.2	6.90 d (8.8)	116.2	6.90 d (8.2)
6'	132.4	8.13 d (8.8)	123.4	7.68 dd (2.2, 8.2)
OMe-7	56.5	3.95 s	56.5	3.85 s
gal				
1	101.0	5.44 d (7.5)	101.4	5.42 d (7.6)
2	76.8	3.93 dd (7.9, 9.8)	76.6	3.96 dd (7.6, 9.5)
3	75.1	3.71 <sup>a</sup>	75.1	3.72 <sup>a</sup>
4	70.6	3.78 dd (0.6, 3.1)	70.7	3.81 dd (1.0, 3.1)
5	74.4	3.70 <sup>a</sup>	74.3	3.70 <sup>a</sup>
6	64.5	4.11 dd (7.6, 11.4)	64.2	4.09 <sup>a</sup>
		4.13 dd (7.6, 11.4)		4.09 <sup>a</sup>
api				
1	109.8	5.45 d (1.6)	110.8	5.46 d (1.3)
2	78.1	4.05 d (1.6)	78.1	4.05 d (1.3)
3	80.9		80.9	
4	75.5	3.72 d (9.8)	75.6	3.70 d (10.3)
		4.07 d (9.8)		4.06 d (10.3)
5	66.2	3.63 d (11.6)	66.3	3.64 d (11.4)
		3.72 d (11.6)		3.75 d (11.4)
glutaroyl				
1	172.2		172.2	
2	46.0	2.35 d (15.1)	46.8	2.35 d (13.9)
		2.42 d (15.1)		2.41 d (13.9)
3	70.6		70.7	
4	46.3	2.36 d (14.9)	46.8	2.25 d (15.4)
		2.46 d (14.9)		2.38 d (15.4)
5	175.6		176.0	
Me-3	27.9	1.13 s	27.7	1.10 s

<sup>a</sup> Multiplicities were unclear due to overlapping with other signals.

methylglutaric acid is rarely found in the flavonoid literature.<sup>9</sup>

Compound **2** was isolated as an amorphous yellow powder. Its UV spectral properties and ESIMS showed that **2** differed from **1** only by an additional hydroxy group at C-3' on the aglycon.<sup>5,6</sup> The HRFABMS (positive-ion mode) of **2** exhibited a pseudomolecular ion peak at  $m/z$  777.1864  $[\text{M} + \text{Na}]^+$  (calcd 777.1854), consistent with a molecular formula of  $\text{C}_{33}\text{H}_{38}\text{O}_{20}$ . Chromatography of the acid hydrolysis products allowed the identification of 7-methylquercetin (rhamnetin), and apiose and galactose. The occurrence in the  $^1\text{H}$  NMR spectrum of an ABM spin system [ $\delta$  7.72 (d,  $J = 2.2$  Hz), 6.90 (d,  $J = 8.2$  Hz), and 7.68 (dd,  $J = 2.2, 8.2$  Hz)] characterized an *ortho*-disubstituted B ring (Table 1). Thus, **2** was identified as rhamnetin-3-*O*-{[3-hydroxy-3-methylglutaroyl(1 $\rightarrow$ 6)] $[\beta$ -D-apiofuranosyl(1 $\rightarrow$ 2)] $[\beta$ -D-galactopyranoside]. According to the literature, rhamnetin is rare in the genus *Astragalus*.<sup>10</sup>

Compound **3** was isolated as an amorphous yellow powder. Its chromatographic behavior and UV spectral properties suggested **3** to be a kaempferol 3-oligoglycoside.<sup>5,6</sup> The HRFABMS (positive-ion mode) of **3** exhibited a pseudomolecular ion peak at  $m/z$  749.1907  $[\text{M} + \text{Na}]^+$  (calcd 749.1905), consistent with the molecular formula

**Table 2.**  $^{13}\text{C}$  and  $^1\text{H}$  NMR Spectral Data of Compounds **3** in  $\text{CD}_3\text{OD}$  and **4** in  $\text{DMSO}-d_6$ 

position	<b>3</b>		<b>4</b>	
	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J/Hz)	$\delta_{\text{C}}$	$\delta_{\text{H}}$ (J/Hz)
aglycon				
2	159.4		156.7	
3	135.8		134.0	
4	179.6		178.1	
5	162.9		161.9	
6	100.0	6.20 d (1.9)	99.4	6.19 d (1.9)
7	166.0		164.8	
8	95.0	6.39 d (1.9)	94.3	6.39 d (1.9)
9	158.5		157.1	
10	105.6		104.7	
1'	122.6		122.0	
2'	132.6	8.10 d (8.8)	116.5	7.60 d (2.2)
3'	116.1	6.88 d (8.8)	145.6	
4'	161.6		149.1	
5'	116.1	6.88 d (8.8)	116.0	6.82 d (8.8)
6'	132.6	8.10 d (8.8)	123.1	7.74 dd (2.2, 8.8)
gal				
1	105.6	5.02 d (7.9)	100.0	5.52 d (7.9)
2	73.0	3.80 dd (7.9, 9.5)	75.5	3.78 dd (7.9, 10.1)
3	74.9	3.56 dd (3.5, 9.5)	74.1	3.62 <sup>a</sup>
4	70.0	3.76 <sup>a</sup>	69.1	3.60 <sup>a</sup>
5	75.5	3.66 <sup>a</sup>	74.0	3.58 <sup>a</sup>
6	68.0	3.48 <sup>a</sup>	65.9	3.28 <sup>a</sup>
		3.70 <sup>a</sup>		3.57 <sup>a</sup>
rha				
1	101.9	4.52 d (1.9)	100.9	4.39 d (1.6)
2	71.7	3.77 dd (1.9, 3.2)	70.5	3.55 <sup>a</sup>
3	82.4	3.55 dd (3.2, 9.5)	81.8	3.38 <sup>a</sup>
4	72.7	3.44 t (9.5)	71.6	3.32 <sup>a</sup>
5	69.5	3.54 <sup>a</sup>	68.7	3.42 <sup>a</sup>
6	18.1	1.16 d (6.0)	18.6	1.06 d (6.0)
xyl				
1	106.5	4.32 d (7.2)	106.0	4.27 d (7.3)
2	74.9	3.23 dd (7.2, 9.1)	74.6	3.05 <sup>a</sup>
3	77.3	3.31 <sup>a</sup>	76.7	3.12 <sup>a</sup>
4	71.1	3.46 m	70.2	3.25 ddd (5.0, 9.1, 13.2)
5	66.9	3.09 dd (10.4, 11.4)	66.4	3.05 <sup>a</sup>
		3.76 <sup>a</sup>		3.66 <sup>a</sup>
api				
1			109.6	5.31 d (1.6)
2			76.9	3.82 <sup>a</sup>
3			80.0	
4			74.8	3.85 <sup>a</sup>
				3.50 <sup>a</sup>
5			65.1	3.39 d (8.9)
				3.47 d (8.9)

<sup>a</sup> Multiplicities were unclear due to overlapping with other signals.

$\text{C}_{32}\text{H}_{38}\text{O}_{19}$ . The ESIMS gave a pseudomolecular ion peak  $[\text{M} + \text{H}]^+$  at  $m/z$  727 and three other ion peaks at  $m/z$  595  $[\text{M} + \text{H} - 132]^+$ , 449  $[\text{M} + \text{H} - 132 - 146]^+$ , and 287  $[\text{M} + \text{H} - 132 - 146 - 162]^+$ , indicating a terminal pentosyl, an intermediate rhamnosyl, and a primary hexosyl. Acid hydrolysis gave kaempferol and three sugars, identified by co-TLC as xylose, rhamnose, and galactose. On the basis of its  $^1\text{H}$  and  $^{13}\text{C}$  NMR data, the triglycoside **3** differed from a previously published kaempferol derivative from *A. caprinus* by the absence of apiose (Table 2).<sup>1</sup> Three anomeric  $^1\text{H}$  NMR signals at  $\delta$  5.02 (d,  $J = 7.9$  Hz), 4.52 (d,  $J = 1.9$  Hz), and 4.32 (d,  $J = 7.2$  Hz) were assigned to three corresponding  $^{13}\text{C}$  NMR signals via HSQC correlations. A HSQC-TOCSY experiment and coupling constant measurements permitted the identification of the three sugar units as  $\beta$ -galactopyranosyl (Gal),  $\alpha$ -rhamnopyranosyl (Rha), and  $\beta$ -xylopyranosyl (Xyl) (Table 2). HMBC experiments showed correlations between Gal H-1 ( $\delta$  5.02) and kaempferol C-3 ( $\delta$  135.8), Rha H-1/C-1 ( $\delta$  4.52/101.9) and Gal C-6/H-6 ( $\delta$  68.0/3.48), and Xyl H-1/C-1 ( $\delta$  4.32/106.5) and Rha C-3/H-3

( $\delta$  82.4/3.55). Consequently, the structure of **3** was established as kaempferol-3-*O*-[ $\beta$ -D-xylopyranosyl(1 $\rightarrow$ 3)]- $\alpha$ -L-rhamnopyranosyl(1 $\rightarrow$ 6)]- $\beta$ -D-galactopyranoside. According to the literature, flavonol tri- or higher glycosides are infrequent in the genus *Astragalus*.<sup>11–13</sup> The present work on *A. caprinus* has resulted in only the second report of a flavonol xyloside in this genus.<sup>1,14</sup>

Compound **4** was isolated as an amorphous yellow powder. Its chromatographic behavior and UV spectral properties suggested this compound to be a quercetin 3-oligoglycoside.<sup>5,6</sup> The HRFABMS (positive-ion mode) of **4** exhibited a pseudomolecular ion peak at  $m/z$  897.2293 [M + Na]<sup>+</sup> (calcd 897.2277), consistent with a molecular formula of C<sub>37</sub>H<sub>46</sub>O<sub>24</sub>. The ESIMS (positive-ion mode) displayed a pseudomolecular ion peak [M + H]<sup>+</sup> at  $m/z$  875 and showed that **4** differed from the previously published kaempferol tetraglycoside from this same plant only by an additional hydroxy group on the aglycon.<sup>1</sup> This was confirmed by acid hydrolysis, which yielded quercetin and the same sugars, namely, galactose, apiose, xylose, and rhamnose. The <sup>1</sup>H and <sup>13</sup>C NMR spectral data showed that **4** is a quercetin tetraglycoside with the same saccharide structure as an already known kaempferol tetraglycoside from this species (Table 2).<sup>1</sup> Four anomeric doublet signals corresponded to four sugar units at  $\delta$  4.27 (1H,  $J = 7.3$  Hz), 4.39 (1H,  $J = 1.6$  Hz), 5.31 (1H,  $J = 1.6$  Hz), and 5.52 (1H,  $J = 7.9$  Hz). A HMBC experiment allowed the glycosidic linkages to be determined between kaempferol C-3 ( $\delta$  134.0) and galactose H-1 ( $\delta$  5.52), Gal C-2 ( $\delta$  75.5) and Api H-1 ( $\delta$  5.31), Gal C-6 ( $\delta$  65.9) and Rha H-1 ( $\delta$  4.39), and Rha C-3 ( $\delta$  81.8) and Xyl H-1 ( $\delta$  4.27). Consequently, the structure of **4** was established as quercetin-3-*O*-{[ $\beta$ -D-xylopyranosyl(1 $\rightarrow$ 3)]- $\alpha$ -L-rhamnopyranosyl(1 $\rightarrow$ 6)]-[ $\beta$ -D-apiofuranosyl(1 $\rightarrow$ 2)]}- $\beta$ -D-galactopyranoside.

## Experimental Section

**General Experimental Procedures.** Optical rotations were measured on a Perkin-Elmer 241 polarimeter. UV and IR spectra were recorded on a Kontron Uvikon-860 and a Perkin-Elmer 681 spectrophotometer, respectively. The 1D and 2D NMR spectra (HSQC, HSQC-TOCSY, HMBC) were performed in CD<sub>3</sub>OD for **1–3** and DMSO-*d*<sub>6</sub> for **4**, using a Bruker DRX 500 NMR spectrometer equipped to allow inverse detection (500 MHz for <sup>1</sup>H and 125 MHz for <sup>13</sup>C). ESIMS data were obtained on a Hewlett-Packard 1100 MSD spectrometer (100 and 150 eV) in the positive-ion mode. HRFABMS were obtained in the positive-ion mode (thioglycerol matrix) on a ZAB2-SEQ instrument. TLC were performed on polyamide [DC6, Macherey-Nagel; mobile phase, H<sub>2</sub>O–MeOH–MeCOEt–2,4-pentanedione (13:3:3:1); system *p*] and cellulose [DC-microcrystalline, Merck, mobile phase, *n*-BuOH–AcOH–H<sub>2</sub>O (4:1:5 upper phase); system *c*]. HPLC was conducted on a Kontron LC set (autosampler 360, pump 322) coupled to a Waters 991 diode array detector on a reversed-phase C<sub>18</sub> Ultrabase (5  $\mu$ m) column (200  $\times$  4.6 mm i.d.), with a linear gradient from 10 to 45% of MeCN in H<sub>2</sub>O with 2% AcOH for 118 min, a flow rate of 0.8 mL min<sup>-1</sup>, and a detection range of 230–410 nm (system *hp*). MPLC separations were performed on a Büchi system equipped with a B-688 pump coupled to a B-687 gradient programmer, using 110  $\times$  15 mm precolumns, 460  $\times$  26 mm columns, filled with the stationary phase C<sub>18</sub> [(40–63  $\mu$ m, Merck), solvent, 50% MeOH in water, flow rate, 10 mL min<sup>-1</sup>; system *rmp*], and 460  $\times$  15 mm columns, filled with the stationary phase Lichroprep-diol [(40–63  $\mu$ m, Merck) gradient of *i*-PrOH in CHCl<sub>3</sub> (linear 5 to 30% in 210 min, flow 6 mL min<sup>-1</sup>; system *sm*)]. Elution was monitored with a Knauer K-2501 UV detector set at 350 nm. Purification was carried out over Sephadex (LH-20, Pharmacia).

**Plant Material.** The leaves of *Astragalus caprinus* were collected during April–May 1998 at Monastir, Tunisia. A

voucher specimen (No. 98/0039) is deposited at the Faculty of Pharmacy, University of Monastir, Tunisia.

**Extraction and Isolation.** Dried, powdered leaves (200 g) were macerated twice with 70% MeOH (500 mL) for 24 h. After partial evaporation the aqueous solution was extracted with *n*-BuOH (3  $\times$  300 mL). The residue from the *n*-BuOH layer was prepurified on a cellulose column (5% EtOH), then separated on a polyamide column (5 to 40% EtOH). The fraction eluted with 5% EtOH was subjected to MPLC (system *rmp*), with the subfraction containing **4** (eluted with 50% MeOH) chromatographed on Sephadex with 50% MeOH to yield pure **4** (26 mg). The fraction eluted from polyamide with 25% EtOH was subjected to MPLC (system *sm*), with the subfraction containing **3** (eluted with 20% *i*-PrOH) purified by preparative TLC (system *p*), then by passage over Sephadex, eluted with MeOH, to yield **3** (13 mg). The fraction eluted from polyamide with 40% EtOH was subjected to MPLC (system *rmp*), with the subfractions containing **1** and **2**, respectively, then submitted to preparative TLC (system *p*, twice) followed by Sephadex eluted with MeOH to afford **1** (5 mg) and **2** (2 mg). Purity was monitored by HPLC (system *hp*). Compounds **1–4** were obtained as pale yellow amorphous powders.

**Rhamnocitrin-3-*O*-{[3-hydroxy-3-methylglutaroyl(1 $\rightarrow$ 6)]-[ $\beta$ -D-apiofuranosyl(1 $\rightarrow$ 2)]}- $\beta$ -D-galactopyranoside (1):** [ $\alpha$ ]<sub>D</sub><sup>20</sup> -81° (*c* 0.14, MeOH); UV  $\lambda_{\max}$  (MeOH) 266 (log  $\epsilon$  4.39), 345 (log  $\epsilon$  4.29) nm; (MeOH + AlCl<sub>3</sub>) 275, 293, 347, 397 nm; (MeOH + AlCl<sub>3</sub> + HCl) 274, 293, 346, 395 nm; (MeOH + NaOAc) 266, 351 nm; (MeOH + NaOAc + H<sub>3</sub>BO<sub>3</sub>) 266, 345 nm; (MeOH + NaOH) 266, 389 nm; (MeOH + NaOH, after 10 min) 266, 389 nm; IR  $\nu_{\max}$  (KBr) 3420, 2940, 1720, 1660, 1595, 1495, 1350, 1215, 1170, 1270, 1085, 1060 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; ESIMS  $m/z$  739 [M + H]<sup>+</sup> (100), 607 [M + H - 132]<sup>+</sup> (60), 301 [M + H - 132 - 306]<sup>+</sup> (65); positive HRFABMS  $m/z$  761.1908 [M + Na]<sup>+</sup> (calcd for C<sub>33</sub>H<sub>38</sub>O<sub>19</sub>Na, 761.1905); TLC *R*<sub>f</sub> 0.76 (system *c*), 0.20 (system *p*); HPLC *R*<sub>t</sub> 78.2 min (system *hp*).

**Rhamnetin-3-*O*-{[3-hydroxy-3-methylglutaroyl(1 $\rightarrow$ 6)]-[ $\beta$ -D-apiofuranosyl(1 $\rightarrow$ 2)]}- $\beta$ -D-galactopyranoside (2):** [ $\alpha$ ]<sub>D</sub><sup>20</sup> -67° (*c* 0.13, MeOH); UV  $\lambda_{\max}$  (MeOH) 256 (log  $\epsilon$  4.40), 267, 354 (log  $\epsilon$  4.28) nm; (MeOH + AlCl<sub>3</sub>) 275, 423 nm; (MeOH + AlCl<sub>3</sub> + HCl) 270, 358 (sh), 270 nm; (MeOH + NaOAc) 258, 357 nm; (MeOH + NaOAc + H<sub>3</sub>BO<sub>3</sub>) 260, 369 nm; (MeOH + NaOH) 264, 393 nm; (MeOH + NaOH, after 10 min) 264, 392 nm; IR  $\nu_{\max}$  (KBr) 3410, 2935, 1715, 1655, 1600, 1500, 1350, 1215, 1270, 1085, 1060 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 1; ESIMS  $m/z$  755 [M + H]<sup>+</sup> (100), 623 [M + H - 132]<sup>+</sup> (20), 317 [M + H - 132 - 306]<sup>+</sup> (35); positive HRFABMS  $m/z$  777.1864 [M + Na]<sup>+</sup> (calcd for C<sub>33</sub>H<sub>38</sub>O<sub>20</sub>Na, 777.1854); TLC *R*<sub>f</sub> 0.66 (system *c*), 0.18 (system *p*); HPLC *R*<sub>t</sub> 70.2 min (system *hp*).

**Kaempferol-3-*O*-[ $\beta$ -D-xylopyranosyl(1 $\rightarrow$ 3)]- $\alpha$ -L-rhamnopyranosyl(1 $\rightarrow$ 6)]- $\beta$ -D-galactopyranoside (3):** [ $\alpha$ ]<sub>D</sub><sup>20</sup> -33.4° (*c* 0.46, MeOH); UV  $\lambda_{\max}$  (MeOH) 266 (log  $\epsilon$  4.36), 349 (log  $\epsilon$  4.30) nm; (MeOH + AlCl<sub>3</sub>) 274, 304, 348, 398 nm; (MeOH + AlCl<sub>3</sub> + HCl) 275, 304, 343, 396 nm; (MeOH + NaOAc) 272, 357 nm; (MeOH + NaOAc + H<sub>3</sub>BO<sub>3</sub>) 266, 339 nm; (MeOH + NaOH) 273, 324, 390 nm; (MeOH + NaOH, after 10 min) 273, 324, 390 nm; IR  $\nu_{\max}$  (KBr) 3400, 2915, 1655, 1605, 1495, 1355, 1205, 1280, 1085, 1060 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 2; ESIMS  $m/z$  727 [M + H]<sup>+</sup> (100), 595 [M + H - 132]<sup>+</sup> (15), 449 [M + H - 132 - 146]<sup>+</sup> (10), 287 [M + H - 132 - 146 - 162]<sup>+</sup> (10); positive HRFABMS  $m/z$  749.1907 [M + Na]<sup>+</sup> (calcd for C<sub>32</sub>H<sub>38</sub>O<sub>19</sub>Na, 749.1905); TLC *R*<sub>f</sub> 0.50 (system *c*), 0.61 (system *p*); HPLC *R*<sub>t</sub> 33.2 min (system *hp*).

**Quercetin-3-*O*-{[ $\beta$ -D-xylopyranosyl(1 $\rightarrow$ 3)]- $\alpha$ -L-rhamnopyranosyl(1 $\rightarrow$ 6)]-[ $\beta$ -D-apiofuranosyl(1 $\rightarrow$ 2)]}- $\beta$ -D-galactopyranoside (4):** [ $\alpha$ ]<sub>D</sub><sup>20</sup> -78.6° (*c* 0.17, MeOH); UV  $\lambda_{\max}$  (MeOH) 255 (log  $\epsilon$  4.41), 267, 356 (log  $\epsilon$  4.32) nm; (MeOH + AlCl<sub>3</sub>) 275, 435 nm; (MeOH + AlCl<sub>3</sub> + HCl) 269, 362, 401 nm; (MeOH + NaOAc) 272, 369 nm; (MeOH + NaOAc + H<sub>3</sub>BO<sub>3</sub>) 260, 374 nm; (MeOH + NaOH) 271, 405 nm; (MeOH + NaOH, after 10 min) 271, 405 nm; IR  $\nu_{\max}$  (KBr) 3380, 2920, 1652, 1605, 1500, 1355, 1205, 1180, 1280, 1130, 1080, 1045 cm<sup>-1</sup>; <sup>1</sup>H and <sup>13</sup>C NMR data, see Table 2; ESIMS  $m/z$  875 [M + H]<sup>+</sup> (100), 743

$[M + H - 132]^+$  (15), 597  $[M + H - 132 - 146]^+$  (10), 465  $[M + H - 132 - 146 - 132]^+$  (10), 303  $[M + H - 132 - 146 - 132 - 162]^+$  (10); positive HRFABMS  $m/z$  897.2293  $[M + Na]^+$  (calcd for  $C_{37}H_{46}O_{24}Na$ , 897.2277); TLC  $R_f$  0.35 (system *c*), 0.83 (system *p*); HPLC  $R_t$  23.8 min (system *hp*).

**Acid Hydrolysis of Compounds 1–4.** Performed using a previously described procedure.<sup>1</sup>

**Acknowledgment.** This work was supported by a grant and a scholarship from the Rhone-Alpes (Fr)-Monastir (Tn) Research Convention.

### References and Notes

- (1) Semmar, N.; Fenet, B.; Lacaille-Dubois, M. A.; Gluchoff-Fiasson, K.; Chemli, R.; Jay, M. *J. Nat. Prod.* **2001**, *64*, 656–658.
- (2) Markham, K. R.; Geiger, H. In *The Flavonoids: Advances in Research since 1986*; Harborne, J. B., Ed.; Chapman & Hall: London, 1994; pp 441–497.
- (3) Bohm, B. A. S. *Introduction to Flavonoids: Chemistry and Biochemistry of Organic Natural Products*, Harwood Academic Publishers: Australia, 1998; Vol. 2.
- (4) Harborne, J. B.; Baxter, H. *The Handbook of the Natural Flavonoids*; Harborne and Baxter, Eds.; Wiley & Son: Chichester, 1999; Vol. 1.
- (5) Mabry, T. J.; Markham, K. R.; Thomas, M. B. *Systematic Identification of Flavonoids*; Springer-Verlag: New York, 1970.
- (6) Markham, K. R. *Techniques of Flavonoid Identification*; Academic Press: London, 1982.
- (7) Agrawal, P. K.; Thakur, R. S.; Bansal, M. C. In *Carbon-13 NMR of Flavonoids*; Agrawal P. K., Ed.; Elsevier: Amsterdam, 1989; pp 150–155.
- (8) Bashir, A.; Hamburger, M.; Hiller, K.; Gupta, M. P.; Krause, E.; Solis, P. N.; Hostettmann, K. *Phytochemistry* **1991**, *30*, 3781–3784.
- (9) Tschan, G. M.; König, G. M.; Wright, A. D.; Sticher, O. *Phytochemistry* **1996**, *41*, 643–646.
- (10) Yasinov, R. K.; Syrovezho, N. V.; Yakovlev, G. P.; Ovcharenko, S. N. *Khim. Prir. Soedin.* **1984**, *4*, 523–524.
- (11) Bedir, E.; Calis, I.; Piacente, S.; Pizza, C.; Khan, I. A. *Chem. Pharm. Bull.* **2000**, *48*, 1994–1995.
- (12) Yahara, S.; Kohhjouma, M.; Kohoda, H. *Phytochemistry* **2000**, *53*, 469–471.
- (13) Alaniya, M. D.; Chkadia, N. F. *Chem. Nat. Compd.* **2000**, *36*, 537.
- (14) Alaniya, M. D.; Komissarenko, N. F.; Kemertelidze, E. P. *Izv. Akad. Nauk Gruz SSR, Ser. Khim.* **1976**, *2*, 31–38.

NP010328L